OSSC Main Written Examination for Regular Teacher 22nd Sep 2023 S2 Tentative Score

Participant ID		
Participant Name		
Test Center Name		
Test Date	22/09/2023	
Test Time	12:30 PM - 2:30 PM	
Subject	TGT Science CBZ	
Marks Obtained		

Section: Pedagogy

Q.1 According to constructivist approach to teaching-learning, the learner is considered

as:

Ans X A. Raw material to be shaped as desired.

🗶 B. Somebody who is not exposed to world around.

X C. An empty vessel to be filled with information.

D. An active individual with rich past experiences.

Question Type: MCQ

Question ID: 4906393604 Option 1 ID: 49063914413 Option 2 ID: 49063914415 Option 3 ID: 49063914414 Option 4 ID: 49063914416

Status : Not Answered

Chosen Option : --Marks : 0

Q.2 Which of the following types of validity is relevant for an achievement test?

Ans X A. Predictive validity

X B. Concurrent validity

C. Content validity

X D. Construct validity

Question Type: MCQ

Question ID: 4906393618 Option 1 ID: 49063914472 Option 2 ID: 49063914471 Option 3 ID: 49063914469 Option 4 ID: 49063914470

Status: Answered

Chosen Option : **C**Marks : **1**

Q.3 Which of the following characteristics of a learner may help the teacher in designing effective teaching-learning conditions?

Ans 🗳 A. Pre-requisite knowledge of the learner

🗶 B. Learner's family background

X C. Good teacher-pupil relationship.

X D. Location of the school building

Question Type: MCQ

Question ID: 4906393606 Option 1 ID: 49063914421 Option 2 ID: 49063914423 Option 3 ID: 49063914422 Option 4 ID: 49063914424 Status: Answered

Chosen Option : A Marks : 1

Q.4 For any test of ability, aptitude, or achievement, which of the following is the most important characteristic?

Ans 🛷 A. Validity

X B. Reliability

X C. Usability

X D. Objectivity

Question Type: MCQ

Question ID: 4906393619 Option 1 ID: 49063914476 Option 2 ID: 49063914474 Option 3 ID: 49063914475

Option 4 ID : 49063914473
Status : Answered

Chosen Option : **B**Marks : -1/4

Q.5 A test used for forecasting an individual's potential for future learning in some specific area, is technically called:

Ans X A. Diagnostic test

X B. Achievement test

X C. Ability test.

D. Aptitude test

Question Type: MCQ

Question ID: 4906393620 Option 1 ID: 49063914477 Option 2 ID: 49063914479

Option 3 ID: 49063914480 Option 4 ID: 49063914478

Status : **Answered**

Chosen Option : **D** Marks : **1**

Q.6 Which of the following is a disadvantage of lecture method of teaching?

Ans X A. Teacher can address large group of learners at the same time.

- X B. It is economical in terms of man, money, and material.
- X C. Its coverage may be further improved by using microphone.
- D. It is one-way communication which does not activate learners.

Question Type: MCQ

Question ID: 4906393609 Option 1 ID: 49063914433 Option 2 ID: 49063914436 Option 3 ID: 49063914434 Option 4 ID: 49063914435 Status: Answered

Chosen Option: D Marks: 1

Q.7 When assessment is done for the purpose of certification, it is an example of:

A. Summative assessment Ans

X B. Prognostic assessment.

X C. Formative assessment

X D. Internal assessment

Question Type: MCQ

Question ID: 4906393614 Option 1 ID: 49063914455 Option 2 ID: 49063914456 Option 3 ID: 49063914454 Option 4 ID: 49063914453 Status: Answered

Chosen Option: A Marks: 1

Q.8 Which of the following types of testing may be categorised as assessment of learning?

Ans X A. Ability testing

🗶 B. Diagnostic testing

C. Achievement testing

X D. Aptitude testing

Question Type: MCQ

Question ID: 4906393613 Option 1 ID: 49063914452 Option 2 ID: 49063914449 Option 3 ID: 49063914450 Option 4 ID: 49063914451

Status : Answered

Chosen Option : C

Marks: 1

Q.9 Which of the following is NOT the purpose of formative assessment?

Ans X A. To motivate learners to work harder.

- X B. To monitor learner's ongoing progress in learning.
- X C. To provide feedback for improvement of teaching.
- D. To prepare an end-term report card of learning.

Question Type: MCQ

Question ID: 4906393612
Option 1 ID: 49063914447
Option 2 ID: 49063914448
Option 3 ID: 49063914446
Option 4 ID: 49063914445
Status: Answered

Chosen Option : **D** Marks : **1**

Q.10 A rubric developed as a part of assessment process acts as a:

Ans X A. Measuring tool.

X B. Seating plan

X C. Optical scanner.

D. Scoring guide

Question Type: MCQ

Question ID: 4906393616 Option 1 ID: 49063914464 Option 2 ID: 49063914462 Option 3 ID: 49063914461 Option 4 ID: 49063914463 Status: Answered

Chosen Option : A Marks : -1/4

Q.11 Which of the following statements describes a characteristic of summative assessment?

Ans X A. It helps the teacher to modify his method of teaching.

X B. It informs the learner of how well he is learning.

C. It assesses learning performance at the end of the term.

X D. It evaluates ongoing learning progress of the learner.

Question Type: MCQ

Question ID : **4906393611** Option 1 ID : **49063914442**

Option 2 ID : 49063914443 Option 3 ID : 49063914444

Option 4 ID : 49063914441

Status : **Answered** Chosen Option : **C**

Marks : 1

Q.12 Which of the following teacher behaviour is conducive to enhancing learners' motivation?

Ans X A. Checking daily homework

X B. Discouraging truancy

C. Reinforcing desirable behaviour

X D. Promoting competition

Question Type: MCQ

Question ID: 4906393607 Option 1 ID: 49063914426 Option 2 ID: 49063914427 Option 3 ID: 49063914428 Option 4 ID: 49063914425

Status : **Answered** Chosen Option : **C**

Marks: 1

Q.13 The focus of constructivist theory of learning is on:

Ans X A. Encouraging individual self-study.

X B. Memory based learning.

C. Interactive and discussion-based learning.

X D. Following teacher's directions in toto.

Question Type: MCQ

Question ID: 4906393603 Option 1 ID: 49063914412 Option 2 ID: 49063914409 Option 3 ID: 49063914410 Option 4 ID: 49063914411

Status : Answered

Chosen Option : A Marks : -1/4

Q.14 Learning is a continuous process of:

Ans X A. Growth and development.

X B. Developing survival skills.

C. Change in behaviour.

X D. Acquiring knowledge.

Question Type: MCQ

Question ID: 4906393602 Option 1 ID: 49063914405 Option 2 ID: 49063914407 Option 3 ID: 49063914408 Option 4 ID: 49063914406

Status : Answered

Chosen Option : **C** Marks : **1**

Q.15 The effectiveness of teaching-learning process is assessed in terms of:

Ans X A. Extent of ICT used by the teacher.

X B. Amount of subject matter taught.

X C. Scores of students in the examination.

D. Learning outcomes relative to objectives.

Question Type: MCQ

Question ID: 4906393601 Option 1 ID: 49063914404 Option 2 ID: 49063914402 Option 3 ID: 49063914401 Option 4 ID: 49063914403 Status: Not Answered

Chosen Option : --Marks : 0

Q.16 Which of the following teaching approaches believes that learner can build knowledge on the foundation of his experiences?

Ans X A. Curriculum-centric approach.

🗶 B. Learner-centric approach

C. Constructivist approach

X D. Teacher-centric approach

Question Type: MCQ

Question ID: 4906393610
Option 1 ID: 49063914440
Option 2 ID: 49063914439
Option 3 ID: 49063914438
Option 4 ID: 49063914437
Status: Answered

Chosen Option : **C** Marks : **1**

Q.17 While selecting teachers for undergraduate teaching, previous work of candidates like book, paper/articles published, and certificates of other accomplishments are examined. This is a kind of:

Ans X A. Formative assessment.

X B. Project work assessment

X C. Summative assessment

D. Portfolio assessment.

Question Type: MCQ

Question ID: 4906393615 Option 1 ID: 49063914457

Option 2 ID : **49063914459** Option 3 ID : **49063914458**

Option 4 ID: 49063914460

Status: Not Answered

Chosen Option : --Marks : 0

Q.18 The first step in the process of development of an achievement test is to:

Ans X A. Prepare a blueprint for the proposed test.

B. List course objectives to be achieved.

X C. Decide the number and type of test-items.

X D. Construct a content-based pool of test-items.

Question Type: MCQ

Question ID: 4906393617 Option 1 ID: 49063914465 Option 2 ID: 49063914467 Option 3 ID: 49063914468 Option 4 ID: 49063914466 Status: Answered

Chosen Option : A

Marks : -1/4

Q.19 Which of the following statements is true for relationship between learning and maturation?

Ans A. Learning-maturation relationship is complex.

🗶 B. Learning and maturation are synonymous.

X C. Learning is not related to maturation.

🗶 D. Learning precedes maturation..

Question Type: MCQ

Question ID: 4906393608 Option 1 ID: 49063914432 Option 2 ID: 49063914430 Option 3 ID: 49063914429

Option 4 ID : 49063914431 Status : Answered

Chosen Option : **D** Marks : -1/4

Q.20 Which of the following techniques of learning emerged out of humanistic theory of Karl Rogers?

Ans X A. Learning through imitation

X B. Learning by conditioning.

C. Learning by self-initiation

X D. Learning by discovery

Question Type: MCQ

Question ID : 4906393605

Option 1 ID: 49063914419 Option 2 ID: 49063914420

Option 3 ID: 49063914418

Option 4 ID: 49063914417

Status : **Answered**

Chosen Option : C Marks : 1

Section: Botany

Q.21 Phototropism is caused in the stimulated stems because of unequal distribution of 🥒 A. Auxin 🗶 B. Cytokinin X C. Ehylene X D. Gibberelin Question Type: MCQ Question ID: 4906393646 Option 1 ID: 49063914581 Option 2 ID: 49063914582 Option 3 ID: 49063914584 Option 4 ID: 49063914583 Status: Answered Chosen Option: A Marks: 1 Q.22 The common method of reproduction in yeast is Ans 🧳 A. Budding X B. Fission X C. Conidia formation X D. Endospore formation Question Type: MCQ Question ID: 4906393652 Option 1 ID: 49063914605 Option 2 ID: 49063914606 Option 3 ID: 49063914607 Option 4 ID: 49063914608 Status: Answered Chosen Option: C Marks: -1/4 Q.23 The culture of anther produces a wide range of X A. Triploid tree X B. Tetraploid tree C. Haploid trees X D. Diploid trees Question Type: MCQ Question ID: 4906393660

Question ID : 4906393660
Option 1 ID : 49063914639
Option 2 ID : 49063914640
Option 3 ID : 49063914637
Option 4 ID : 49063914638
Status : Answered
Chosen Option : D

Marks : -1/4

Q.24 The total number of different unit characters chosen by Mendel for his experiment in Garden Pea were Ans X A. 9 √ B. 7 X C. 8 X D. 6 Question Type: MCQ Question ID: 4906393657 Option 1 ID: 49063914628 Option 2 ID: 49063914626 Option 3 ID: 49063914627 Option 4 ID: 49063914625 Status: Answered Chosen Option : B Marks: 1 Q.25 The source of Oxygen liberated in photosynthesis is Ans **⋉** в. СО₂ X C. Photosynthetic pigments X D. Glucose Question Type: MCQ Question ID: 4906393629 Option 1 ID: 49063914514 Option 2 ID: 49063914513 Option 3 ID: 49063914516 Option 4 ID: 49063914515 Status : Answered Chosen Option : A Marks: 1 Q.26 Mesophytes are a group of plants that grow in/on Ans 🥒 A. Land X B. Sand X C. Rock X D. Water Question Type: MCQ Question ID: 4906393623 Option 1 ID: 49063914491 Option 2 ID: 49063914492 Option 3 ID: 49063914490 Option 4 ID: 49063914489 Status: Not Answered Chosen Option: --Marks : 0

Q.27 Entry of pollen tube in to the ovule through its integument is known as 🗶 A. Chalazogamy X B. Porogamy 🧳 C. Mesogamy X D. Basigamy Question Type: MCQ Question ID: 4906393655 Option 1 ID: 49063914619 Option 2 ID: 49063914617 Option 3 ID: 49063914618 Option 4 ID: 49063914620 Status: Answered Chosen Option : C Marks: 1 Q.28 Ethylene is a Ans X A. Solid hormone B. Gaseous hormone X C. Semisolid hormone 🗶 D. Liquid hormone Question Type : \mathbf{MCQ} Question ID: 4906393648 Option 1 ID: 49063914589 Option 2 ID: 49063914592 Option 3 ID: 49063914591 Option 4 ID: 49063914590 Status: Answered Chosen Option: B Marks: 1 Q.29 Anemophily is linked to pollination by X A. Water X B. Insects X C. Animals D. Wind Question Type: MCQ Question ID: 4906393654 Option 1 ID: 49063914614 Option 2 ID: 49063914616 Option 3 ID: 49063914615 Option 4 ID: 49063914613 Status: Answered Chosen Option : D Marks: 1

Q.30 Ustilago, the causative agent of smut disease, spreads rapidly fast in

Ans 🗳 A. Dry weather

X B. Damp weather

X C. Wet weather

X D. Cloudy weather

Question Type: MCQ

Question ID: 4906393665
Option 1 ID: 49063914660
Option 2 ID: 49063914657
Option 3 ID: 49063914659
Option 4 ID: 49063914658
Status: Answered

Chosen Option : **C**Marks : -1/4

Q.31 Virus free plants are produced from the tissue culture of

Ans X A. Intercalary meristem

B. Apical meristem

🗶 C. Lateral meristems

X D. Pith cell

Question Type : \mathbf{MCQ}

Question ID: 4906393659
Option 1 ID: 49063914634
Option 2 ID: 49063914633
Option 3 ID: 49063914636
Option 4 ID: 49063914635
Status: Answered

Chosen Option : **B** Marks : **1**

Q.32 C4 plants are adopted

Ans X A. Humid climate

X B. Temperate climate

C. Hot and dry climate

X D. Cold and dry climate

Question Type: MCQ

Question ID : 4906393628 Option 1 ID : 49063914509 Option 2 ID : 49063914510 Option 3 ID : 49063914512 Option 4 ID : 49063914511 Status : Answered

Chosen Option : C Marks : 1

Q.33 Heart wood in plant is a

Ans X A. Growing tissue

🗶 C. Living tissue

X D. Decomposed tissue

Question Type: MCQ

Question ID: 4906393651
Option 1 ID: 49063914602
Option 2 ID: 49063914604
Option 3 ID: 49063914601
Option 4 ID: 49063914603
Status: Not Answered

Chosen Option : --Marks : 0

Q.34 The hormone that induces cell division is

Ans X A. Gibberelin

X B. Auxin

🧳 C. Cytokinin

🗶 D. Ehylene

Question Type: MCQ

Question ID: 4906393647 Option 1 ID: 49063914587 Option 2 ID: 49063914585 Option 3 ID: 49063914586 Option 4 ID: 49063914588 Status: Answered

Chosen Option : C Marks : 1

Q.35 Conjoined collateral closed vascular bundles are seen in

Ans 🛷 A. Monocot Stem

X B. Dicot root

X C. Monocot root

X D. Dicot Stem

Question Type: MCQ

Question ID : 4906393643 Option 1 ID : 49063914570 Option 2 ID : 49063914571 Option 3 ID : 49063914572 Option 4 ID : 49063914569 Status : Answered

Chosen Option : A Marks : 1

Q.36 The parenchymatous cells that give buoyancy to floating plants are called as

Ans 🗙 A. Chlorenchyma

X B. Collenchyma

🗶 C. Prosenchyma

D. Aerenchyma

Question Type: MCQ

Question ID: 4906393640
Option 1 ID: 49063914557
Option 2 ID: 49063914559
Option 3 ID: 49063914560
Option 4 ID: 49063914558
Status: Answered

Chosen Option : **D** Marks : **1**

Q.37 Meristematic tissues are characterized by

Ans X A. Thick cell walls with no intracellular space

X B. Thick cell wall with intracellular space

X C. Thin cell wall with intracellular space

 $\ensuremath{\mathscr{Q}}$ D. Thin cell walls with no intercellular space

Question Type: MCQ

Question ID: 4906393639
Option 1 ID: 49063914555
Option 2 ID: 49063914556
Option 3 ID: 49063914554
Option 4 ID: 49063914553
Status: Answered

Chosen Option : **C**Marks : -1/4

Q.38 Asexual reproduction in Ascomycetes takes place commonly through

Ans X A. Zoospores

🧳 B. Conidia

X C. Endospores

X D. Akinites

Question Type: MCQ

Question ID : 4906393653 Option 1 ID : 49063914609 Option 2 ID : 49063914610 Option 3 ID : 49063914612 Option 4 ID : 49063914611 Status : Answered

Chosen Option : A Marks : -1/4

Q.39 Cyclic photophosphorylation produces

Ans

XAATP, NADPH2 and O2

Ø B. AT P

X c. NADPH₂

X D. ATP and NADPH₂

Question Type: MCQ

Question ID: 4906393624 Option 1 ID: 49063914496 Option 2 ID: 49063914494 Option 3 ID: 49063914495 Option 4 ID: 49063914493

Status : Answered

Chosen Option : **B** Marks : **1**

Q.40 The dehybrid cross ratio of 9:3:3:1 represents the

Ans X A. Genotypic ratio

🗶 B. Both genotypic and phenotypic ratios

X C. Allelic Ratio

🧳 D. Phenotypic ratio

Question Type: MCQ

Question ID: 4906393658 Option 1 ID: 49063914629 Option 2 ID: 49063914631 Option 3 ID: 49063914632

Option 4 ID : **49063914630**Status : **Answered**

Chosen Option : **D** Marks : **1**

Q.41 The fungus Phytophthora influenza is the causal agent of

Ans X A. Red blight of potato

X B. Black blight of potato

X C. Early blight of potato

D. Late blight of potato

Question Type : \mathbf{MCQ}

Question ID: 4906393662 Option 1 ID: 49063914646 Option 2 ID: 49063914647

Option 3 ID: 49063914645 Option 4 ID: 49063914648

Status : Answered

Chosen Option : **D** Marks : **1**

Q.42 The alternative (second) host associated with black stem rust of wheat is

Ans X A. Barley

B. Barberry

X C. Wheat

X D. Maize

Question Type: MCQ

Question ID: 4906393663
Option 1 ID: 49063914650
Option 2 ID: 49063914651
Option 3 ID: 49063914649
Option 4 ID: 49063914652
Status: Answered

Chosen Option : **B** Marks : **1**

Q.43 During the dark reaction of photosynthesis

Ans

X A. Water splits

x B. Chlorophyll is activated

× c. 6C sugar breaks down to 3C sugars

√ D. CO₂ is reduced to Organic compounds

Question Type: MCQ

Question ID: 4906393630 Option 1 ID: 49063914517 Option 2 ID: 49063914520

Option 3 ID: 49063914519 Option 4 ID: 49063914518

Status : Answered

Chosen Option : ${\bf D}$

Marks : 1

Q.44 The photoperiodic behavior of plants is mediated by a pigment known as

Ans X A. Florigen

B. Phytochrome

X C. Ferrochrome

🗶 D. Cytochrome

Question Type: MCQ

Question ID: 4906393642

Option 1 ID: 49063914568

Option 2 ID: 49063914565

Option 3 ID: 49063914567

Option 4 ID: 49063914566

Status : **Answered**

Chosen Option : B

Marks : 1

Q.45 The base metal in the chlorophyll molecule is Ans 🗶 A. Iron X B. Zinc X C. Nickel D. Magnesium Question Type: MCQ Question ID: 4906393621 Option 1 ID: 49063914482 Option 2 ID: 49063914484 Option 3 ID: 49063914481 Option 4 ID: 49063914483 Status: Answered Chosen Option : D Marks: 1 Q.46 Which among the following does not include phloem? X A. Companion cells X B. Sieve tubes 🗶 C. Phloem parenchyma D. Wood fibers Question Type : \mathbf{MCQ} Question ID: 4906393641 Option 1 ID: 49063914562 Option 2 ID: 49063914561 Option 3 ID: 49063914563 Option 4 ID: 49063914564 Status: Answered Chosen Option: D Marks: 1 Q.47 The law of segregation is linked to A. Meiosis X B. Amitosis X C. Promitosis X D. Mitosis Question Type: MCQ Question ID: 4906393661 Option 1 ID: 49063914642 Option 2 ID: 49063914643 Option 3 ID: 49063914644 Option 4 ID: 49063914641 Status: Answered Chosen Option : A Marks: 1

Q.48 Coraloid roots are seen with Ans 💢 A. Pinus X B. Gnetum X C. Psilotum D. Cycas Question Type: MCQ Question ID: 4906393627 Option 1 ID: 49063914506 Option 2 ID: 49063914508 Option 3 ID: 49063914507 Option 4 ID: 49063914505 Status: Answered Chosen Option : D Marks: 1 Q.49 Carotenoid pigments protects the plant from X A. Dessication X B. Photorespiration C. Photooxidation 🗶 D. Photosynthesis Question Type : \mathbf{MCQ} Question ID: 4906393634 Option 1 ID: 49063914535 Option 2 ID: 49063914536 Option 3 ID: 49063914533 Option 4 ID: 49063914534 Status: Answered Chosen Option: C Marks: 1 Q.50 Ribbon-shaped chloroplasts can be seen in X A. Chlorella X B. Chlamydomonas X C. Cladophora 🧳 D. Spirogyra Question Type: MCQ Question ID: 4906393631 Option 1 ID: 49063914522 Option 2 ID: 49063914521 Option 3 ID: 49063914523 Option 4 ID: 49063914524 Status: Not Answered Chosen Option: --Marks: 0

Q.51 Elaters seen in bryophytes are responsible

Ans X A. Spore multiplication

X B. Spore development

X C. Spore germination

D. Spore dispersal

Question Type: MCQ

Question ID: 4906393635 Option 1 ID: 49063914538 Option 2 ID: 49063914537 Option 3 ID: 49063914540 Option 4 ID: 49063914539 Status: Not Answered

Chosen Option : --

Marks : 0

$\mathbf{Q.52}$ Tropical plants such as sugarcane show high efficiency for fixation of CO_2 due to

Ans 🧳 A. Hatch Slack Cycle

🗶 B. Calvin Cycle

X C. TCA cycle

X D. EMP pathway.

Question Type: MCQ

Question ID: 4906393636 Option 1 ID: 49063914543 Option 2 ID: 49063914541 Option 3 ID: 49063914542 Option 4 ID: 49063914544

Status : Answered Chosen Option : A Marks : 1

Q.53 Bicolateral vascular bundles are seen with

Ans X A. Rice

🤣 B. Cucurbita

X C. Tobacco

🗶 D. Betel leaf

Question Type: MCQ

Question ID: 4906393650 Option 1 ID: 49063914598 Option 2 ID: 49063914597 Option 3 ID: 49063914600 Option 4 ID: 49063914599

Status: Answered

Chosen Option : **B** Marks : **1**

Q.54 Gemma cups can be seen in X A. Anthoceros X C. Riccia X D. Funaria Question Type: MCQ Question ID: 4906393622 Option 1 ID: 49063914487 Option 2 ID: 49063914486 Option 3 ID: 49063914485 Option 4 ID: 49063914488 Status: Answered Chosen Option : B Marks: 1 Q.55 Protonema appears with members 🗶 A. Hepaticopsida X B. Anthocerotopsida 🧳 C. Bryopsida 🗶 D. Pteropsida Question Type : \mathbf{MCQ} Question ID: 4906393632 Option 1 ID: 49063914525 Option 2 ID: 49063914526 Option 3 ID: 49063914527 Option 4 ID: 49063914528 Status: Answered Chosen Option: C Marks: 1 Q.56 The supplementary factor is represented by phenotypic ratio X A. 9:7 √ B. 9:3:4 X C. 9:3:3:1 X D. 12:3:1 Question Type: MCQ Question ID: 4906393664 Option 1 ID: 49063914653 Option 2 ID: 49063914654 Option 3 ID: 49063914656 Option 4 ID: 49063914655 Status: Answered Chosen Option : A

Marks : -1/4

Q.57 Endosperm in gymnosperm is 🥒 A. Haploid X B. Triploid X C. Diploid X D. Tetraploid Question Type: MCQ Question ID: 4906393626 Option 1 ID: 49063914501 Option 2 ID: 49063914503 Option 3 ID: 49063914502 Option 4 ID: 49063914504 Status: Answered Chosen Option : C Marks : -1/4 Q.58 Internode elongation is primarily associated with Ans A. Gibberellins X B. Cytokinin X C. Abscisic acid X D. Auxins Question Type : \mathbf{MCQ}

Question Type: MCQ
Question ID: 4906393649
Option 1 ID: 49063914595
Option 2 ID: 49063914594
Option 3 ID: 49063914596
Option 4 ID: 49063914593
Status: Answered
Chosen Option: A

Marks : 1

Q.59 Casperian strips are associated with

Ans X A. Medullary rays

X B. Vascular bundles

C. Endodermis

X D. Pericycle

Question Type: MCQ

Question ID : 4906393644
Option 1 ID : 49063914576
Option 2 ID : 49063914575
Option 3 ID : 49063914573
Option 4 ID : 49063914574
Status : Answered

Chosen Option : C Marks : 1

Q.60 Mucoris a Ans 🗶 A. Parasite B. Saprophyte X C. Epiphyte X D. Semi-parasite Question Type: MCQ Question ID: 4906393625 Option 1 ID: 49063914497 Option 2 ID: 49063914498 Option 3 ID: 49063914500 Option 4 ID: 49063914499 Status: Not Answered Chosen Option: --Marks: 0 Q.61 The elements associated with photolysis of water in photosynthesis are X A. Na and K X B. Mo and Cl X C. B and CI 🥒 D. Mn and Cl Question Type : \mathbf{MCQ} Question ID: 4906393637 Option 1 ID: 49063914545 Option 2 ID: 49063914547 Option 3 ID: 49063914546 Option 4 ID: 49063914548 Status: Answered Chosen Option: D Marks: 1 Q.62 The substrate for photorespiration is X A. Serine C. OAA X D. Malic acid Question Type: MCQ Question ID: 4906393633 Option 1 ID: 49063914532 Option 2 ID: 49063914529 Option 3 ID: 49063914531 Option 4 ID: 49063914530 Status: Answered Chosen Option : C Marks : -1/4

Q.63 Triple fusion in angiosperms results in the formation of

Ans X A. Zygote

X B. Oospore

🧳 C. Endosperm

X D. Oosphere

Question Type: MCQ

Question ID: 4906393656
Option 1 ID: 49063914621
Option 2 ID: 49063914623
Option 3 ID: 49063914624
Option 4 ID: 49063914622
Status: Answered

Chosen Option : C Marks : 1

Q.64 Cork cambium is associated with the second growth of stems in

Ans 🧳 A. Extra stelar Area

X B. Intra stelar Area

X C. Stelar Area

X D. Non-Stelar Area

Question Type: MCQ

Question ID: 4906393645
Option 1 ID: 49063914579
Option 2 ID: 49063914580
Option 3 ID: 49063914577
Option 4 ID: 49063914578
Status: Not Answered

Chosen Option : --Marks : 0

Q.65 The fungus responsible for causing the Smut disease belongs to

Ans X A. Deuteromycetes

X C. Ascomycetes

X D. Phycomycetes

Question Type: MCQ

Question ID: 4906393638 Option 1 ID: 49063914552 Option 2 ID: 49063914551 Option 3 ID: 49063914549 Option 4 ID: 49063914550 Status: Answered

Chosen Option : **B** Marks : **1**

Section: Zoology

Q.66 The experimental material used by Mendel for his study in genetics is Ans 🗶 A. Frog X B. Cockroach 🗶 C. Drosophila D. Garden Pea Question Type: MCQ Question ID: 4906393681 Option 1 ID: 49063914724 Option 2 ID: 49063914723 Option 3 ID: 49063914721 Option 4 ID: 49063914722 Status: Answered Chosen Option : D Marks: 1 Q.67 The Vitamin that is synthesized by Bacteria inside the gut is Ans √ A. K **X** B. C X C. D X D. B Question Type : \mathbf{MCQ} Question ID: 4906393687 Option 1 ID: 49063914748 Option 2 ID: 49063914746 Option 3 ID: 49063914747 Option 4 ID: 49063914745 Status: Answered Chosen Option: D Marks: -1/4 Q.68 The nerve centre for hunger and thirst is present in X A. Pons X B. Medulla C. Hypothalamus X D. Thalamus Question Type: MCQ Question ID: 4906393673 Option 1 ID: 49063914690 Option 2 ID: 49063914691 Option 3 ID: 49063914689 Option 4 ID: 49063914692 Status: Answered Chosen Option : B Marks : -1/4

Q.69 Kaziranga National park is linked to the conservation of: A. One horned Rhinos X B. Deers X C. Peacocks X D. Elephants Question Type: MCQ Question ID: 4906393706 Option 1 ID: 49063914822 Option 2 ID: 49063914824 Option 3 ID: 49063914821 Option 4 ID: 49063914823 Status: Answered Chosen Option: A Marks: 1 Q.70 If the coding DNA strand has got the nitrogenous base sequence of (AATTCAAT), the corresponding mRNA will have the sequence of Ans X B. UUTTUGGC X C. AAGTTACC X D. TTAAGTTA Question Type: MCQ Question ID: 4906393683 Option 1 ID: 49063914729 Option 2 ID: 49063914732 Option 3 ID: 49063914731 Option 4 ID: 49063914730 Status: Not Answered Chosen Option: --Marks: 0 Q.71 The number of ${\rm CO_2}$ molecules released during complete oxidation of a single molecule of Glucose is/are 🥒 A. Six Ans X B. Three X C. One X D. Five Question Type: MCQ Question ID: 4906393677 Option 1 ID: 49063914708 Option 2 ID: 49063914707 Option 3 ID: 49063914705 Option 4 ID: 49063914706 Status : Answered Chosen Option: A Marks: 1

Q.72 The chemical formed during anaerobic respiration responsible for Muscle Cramps is

Ans X A. Glutamic acid

B. Lactic acid

X C. Carboxylic acid

X D. Pyruvic acid

Question Type: MCQ

Question ID: 4906393678
Option 1 ID: 49063914711
Option 2 ID: 49063914710
Option 3 ID: 49063914712
Option 4 ID: 49063914709
Status: Answered

Chosen Option : B Marks : 1

Q.73 The cells named podocytes occur in

Ans 🗳 A. Inner wall of Bowman's capsule

X B. Outer wall of Bowman's capsule

🗶 C. Upper wall of Bowman's capsule

🗶 D. Large Intestine

Question Type: MCQ

Question ID: 4906393670
Option 1 ID: 49063914677
Option 2 ID: 49063914678
Option 3 ID: 49063914679
Option 4 ID: 49063914680
Status: Answered

Chosen Option : A Marks : 1

Q.74 The common feature among amylase, rennin, and trypsin is:

Ans X A. All are produced in Stomach

B. All are proteins

X C. All are proteolytic enzymes

X D. All act at a lower pH

Question Type: MCQ

Question ID: 4906393686
Option 1 ID: 49063914743
Option 2 ID: 49063914741
Option 3 ID: 49063914742
Option 4 ID: 49063914744
Status: Answered

Chosen Option : **B** Marks : **1**

Q.75 Crossing over takes place at the sub-stage of: Ans 🗙 A. Diakinesis X B. Zygotene X C. Leptotene D. Pachytene Question Type: MCQ Question ID: 4906393676 Option 1 ID: 49063914704 Option 2 ID: 49063914702 Option 3 ID: 49063914701 Option 4 ID: 49063914703 Status: Answered Chosen Option : D Marks: 1 Q.76 Ultra-filteration occurs due to Ans X A. Circulating blood X B. Secretion C. Glomerular hydrostatic pressure X D. Osmotic concentration Question Type : \mathbf{MCQ} Question ID: 4906393668 Option 1 ID: 49063914671 Option 2 ID: 49063914672 Option 3 ID: 49063914670 Option 4 ID: 49063914669 Status: Answered Chosen Option: C Marks: 1 Q.77 Deforestation generally decreases due to: 🕢 A. Rainfall X B. Drought X C. Soil erosion X D. Global warming Question Type: MCQ Question ID: 4906393703 Option 1 ID: 49063914810 Option 2 ID: 49063914809 Option 3 ID: 49063914812

Option 4 ID : **49063914811**Status : **Answered**

Chosen Option : A Marks : 1

Q.78 Animal cells are connected with each other by:

Ans 🚀 A. Desmosomes

X B. Cell membrane

X C. Plasmodesmata

X D. Plasma membrane

Question Type: MCQ

Question ID: 4906393684
Option 1 ID: 49063914735
Option 2 ID: 49063914733
Option 3 ID: 49063914736
Option 4 ID: 49063914734
Status: Answered

Chosen Option : **D**Marks : -1/4

Q.79 Ecological Pyramids are also called as:

Ans 🗳 A. Ectonian Pyramids

X B. Pyramid of Number

X C. Pyramid of Energy

X D. Pyramid of Biomass

Question Type: MCQ

Question ID: 4906393699
Option 1 ID: 49063914795
Option 2 ID: 49063914793
Option 3 ID: 49063914794
Option 4 ID: 49063914796
Status: Answered

Chosen Option : A Marks : 1

Q.80 Which among the following is not found in case of animal cell?

Ans X A. Cell membrane

X B. Mitochondria

X C. Cytoplasm

🥒 D. Cell wall

Question Type: MCQ

Question ID : 4906393669 Option 1 ID : 49063914676 Option 2 ID : 49063914674 Option 3 ID : 49063914673 Option 4 ID : 49063914675 Status : Answered

Chosen Option : **D** Marks : **1**

Q.81 ADH takes part in

Ans X A. NA+ reabsorption

B. water retention in Nephron

X C. Reducing Urea formation

X D. Absorption of Water from Urine

Question Type: MCQ

Question ID: 4906393671
Option 1 ID: 49063914682
Option 2 ID: 49063914681
Option 3 ID: 49063914683
Option 4 ID: 49063914684
Status: Answered

Chosen Option : **B** Marks : **1**

Q.82 The major gases concerned with Acid rain are

Ans

X A. O3

 $\mathscr{D}^{\text{B.}}$ SO₂ and NO₂

X c. NH4

X D. CO

Question Type : \boldsymbol{MCQ}

Question ID: 4906393708 Option 1 ID: 49063914829 Option 2 ID: 49063914831 Option 3 ID: 49063914832

Option 4 ID : 49063914830 Status : Answered

Chosen Option : B

Marks: 1

Q.83 Haemoglobin remains dissolved in Blood plasma of

Ans 🗶 A. Frog

🗶 B. Hydra

C. Earthworm

X D. Man

Question Type : \mathbf{MCQ}

Question ID: 4906393690

Option 1 ID: 49063914759

Option 2 ID: 49063914760

Option 3 ID: 49063914758

Option 4 ID: 49063914757

Status : **Answered** Chosen Option : **D**

Marks : -1/4

Q.84 In liver, synthesis of Urea takes place by Ans X A. Nitrogen cycle

X B. Kreb's cycle

C. Ornithine cycle

X D. Glycolysis

Question Type: MCQ

Question ID: 4906393667 Option 1 ID: 49063914666 Option 2 ID: 49063914667 Option 3 ID: 49063914665 Option 4 ID: 49063914668 Status: Answered

Chosen Option : C Marks: 1

Q.85 Which of the following, is represented by an Autotroph?

X A. Mushroom

B. Rice plant

C. Mucor

🗶 D. Rhizopus

Question Type : \mathbf{MCQ}

Question ID: 4906393698 Option 1 ID: 49063914790 Option 2 ID: 49063914792 Option 3 ID: 49063914789 Option 4 ID: 49063914791 Status: Answered

Chosen Option: B Marks: 1

Q.86 Exchange of segments between non-homologous chromosomes is known as

X A. Inversion

X B. Transduction

X C. Transversion

D. Translocation

Question Type: MCQ

Question ID: 4906393679 Option 1 ID: 49063914716 Option 2 ID: 49063914715 Option 3 ID: 49063914713 Option 4 ID: 49063914714 Status: Answered

Chosen Option : C Marks : -1/4

Q.87 The largest part of Brain is Ans X A. Medulla oblongata X B. Cerebellum X C. Olfactory lobe D. Cerebrum Question Type: MCQ Question ID: 4906393672 Option 1 ID: 49063914686 Option 2 ID: 49063914688 Option 3 ID: 49063914685 Option 4 ID: 49063914687 Status: Answered Chosen Option : D Marks: 1 Q.88 Oxyntic cells of fundic glands secret Ans X A. Pepsin X C. Renin 🗶 D. Mucus Question Type : \mathbf{MCQ} Question ID: 4906393688 Option 1 ID: 49063914750 Option 2 ID: 49063914751 Option 3 ID: 49063914752 Option 4 ID: 49063914749 Status: Not Answered Chosen Option: --Marks: 0 Q.89 Which gas is not associated with respiration of animal? XA H2O Ø B. M n **x** c. O₂ X D. CO₂

Question Type : MCQ
Question ID : 4906393702
Option 1 ID : 49063914807
Option 2 ID : 49063914808
Option 3 ID : 49063914806
Option 4 ID : 49063914805
Status : Answered

Chosen Option : **B** Marks : **1**

Q.90 The Origin of Species by Natural Selection was authored by:

Ans X A. George Mendel

B. Charles Darwin

X C. Aristottle

X D. Jean B. Lamarck

Question Type: MCQ

Question ID: 4906393707 Option 1 ID: 49063914828 Option 2 ID: 49063914825 Option 3 ID: 49063914827 Option 4 ID: 49063914826 Status: Answered

Chosen Option : **B**Marks : **1**

Q.91 Five kingdom concept of classification was proposed by:

Ans X A. Carolus Linnaeus

B. Robert Whittaker

X C. Theophrastus

X D. Aristotle

Question Type : \mathbf{MCQ}

Question ID: 4906393709
Option 1 ID: 49063914833
Option 2 ID: 49063914834
Option 3 ID: 49063914835
Option 4 ID: 49063914836
Status: Answered

Chosen Option : **B** Marks : **1**

Q.92 Chromatin is composed of

Ans 🛷 A. DNA, RNA and Proteins

X B. DNA and RNA

X C. RNA and Protein

X D. DNA and Protein

Question Type: MCQ

Question ID : 4906393682 Option 1 ID : 49063914728 Option 2 ID : 49063914725 Option 3 ID : 49063914727 Option 4 ID : 49063914726 Status : Answered

Chosen Option : **D**Marks : -1/4

Q.93 The type of fertilization seen in Frog is: 🦪 A. External X B. Tubular X C. Internal X D. Metamorphic Question Type: MCQ Question ID: 4906393705 Option 1 ID: 49063914818 Option 2 ID: 49063914819 Option 3 ID: 49063914817 Option 4 ID: 49063914820 Status: Answered Chosen Option : A Marks: 1 Q.94 Soil erosion can be prevented by: X A. Over grazing X B. Clearing vegetation C. Afforestation 🗶 D. Deforestation Question Type : \mathbf{MCQ} Question ID: 4906393704 Option 1 ID: 49063914813 Option 2 ID: 49063914814 Option 3 ID: 49063914815 Option 4 ID: 49063914816 Status: Answered Chosen Option: C Marks: 1 Q.95 The Olfactory lobes control the sense of X A. Sight X B. Hearing X C. Taste ⊘ D. Smell Question Type: MCQ Question ID: 4906393675 Option 1 ID: 49063914700 Option 2 ID: 49063914698 Option 3 ID: 49063914697 Option 4 ID: 49063914699 Status: Answered Chosen Option : D Marks: 1

Q.96 Human heart is Ans X A. Progenic X B. Neurogenic 🧳 C. Myogenic X D. Eugenic Question Type: MCQ Question ID: 4906393693 Option 1 ID: 49063914771 Option 2 ID: 49063914770 Option 3 ID: 49063914769 Option 4 ID: 49063914772 Status: Answered Chosen Option : C Marks: 1 Maximum contribution of O2 is from X A. Herbs 🗶 B. Grass lands X C. Dense Forests D. Phytoplankton Question Type : \mathbf{MCQ} Question ID: 4906393696 Option 1 ID: 49063914782 Option 2 ID: 49063914784 Option 3 ID: 49063914783 Option 4 ID: 49063914781 Status : Answered Chosen Option : D Marks: 1 Q.98 The element required for Blood coagulation is Ans X A. Na X B. Mn 🥒 C. Ca X D. K Question Type: MCQ Question ID: 4906393691 Option 1 ID: 49063914761 Option 2 ID: 49063914764 Option 3 ID: 49063914762 Option 4 ID: 49063914763 Status : Answered Chosen Option : C Marks: 1

Q.99 How animal cell is different from plant cell?

Ans X A. Presence of Cytoplasm

X B. Presence of Cell membrane

X C. Presence of Mitochondria

D. Presence of Centrosome

Question Type: MCQ

Question ID : 4906393694 Option 1 ID : 49063914776 Option 2 ID : 49063914773 Option 3 ID : 49063914774 Option 4 ID : 49063914775 Status : Answered

Chosen Option : **D** Marks : **1**

Q.100 The blood does not coagulate inside our body because of

Ans X A. NaCL in the blood

X B. Sodium Oxalate in blood

C. Heparin in blood

X D. Blood Pressure

Question Type: MCQ

Question ID: 4906393692
Option 1 ID: 49063914765
Option 2 ID: 49063914767
Option 3 ID: 49063914768
Option 4 ID: 49063914766
Status: Answered

Chosen Option : C Marks : 1

Q.101 The Mitotic stage at which the chromosomes are seen at the shortest and thickest state is:

Ans 🗶 A. Anaphase

🗶 B. Telophase

C. Metaphase

X D. Prophase

Question Type : MCQ

Question ID: 4906393685 Option 1 ID: 49063914739 Option 2 ID: 49063914740 Option 3 ID: 49063914738 Option 4 ID: 49063914737

Status : Answered

Chosen Option : **C** Marks : **1**

Q.102 Which organelle is known as the power house of cell?

Ans X A. Golgi bodies

X B. Chlorophyll

C. Mitochondria

X D. Ribosome

Question Type: MCQ

Question ID: 4906393697 Option 1 ID: 49063914788 Option 2 ID: 49063914785 Option 3 ID: 49063914786 Option 4 ID: 49063914787 Status: Answered

Chosen Option : C Marks : 1

Q.103 The coordination of Various parts, in the human body, is achieved through

Ans 🧳 A. Nervous system

X B. Endocrine system

🗶 C. Muscular system

🗶 D. Reproductive system

Question Type: MCQ

Question ID: 4906393674
Option 1 ID: 49063914693
Option 2 ID: 49063914695
Option 3 ID: 49063914694
Option 4 ID: 49063914696
Status: Answered

Chosen Option : A Marks : 1

Q.104 The morphological and physiological unit of Vertebrate kidney is

Ans 🛷 A. Uniferous tubule

X B. Ureter

X C. Nephridium

X D. Seminiferous tubule

Question Type: MCQ

Question ID : 4906393666
Option 1 ID : 49063914664
Option 2 ID : 49063914661
Option 3 ID : 49063914662
Option 4 ID : 49063914663
Status : Answered

Chosen Option : A Marks : 1

Q.105 A heritable condition of possessing more than two complete sets of Chromosomes is known as Ans 🗶 A. Diploidy X B. Aneuploidy X C. Monoploidy 🤣 D. Polyploidy Question Type: MCQ Question ID: 4906393680 Option 1 ID: 49063914720 Option 2 ID: 49063914717 Option 3 ID: 49063914719 Option 4 ID: 49063914718 Status : Answered Chosen Option : D Marks: 1 Q.106 The blood of human being gets purified in Ans X A. Heart X B. Skin X C. Liver 🧳 D. Lungs Question Type: MCQ Question ID: 4906393689 Option 1 ID: 49063914754 Option 2 ID: 49063914755 Option 3 ID: 49063914756 Option 4 ID: 49063914753 Status : Answered Chosen Option : D Marks: 1 Q.107 The source of energy in an ecosystem is: Ans X A. ATP X B. ADP C. DNA D. Sunlight Question Type: MCQ Question ID: 4906393695 Option 1 ID: 49063914777 Option 2 ID: 49063914778 Option 3 ID: 49063914780 Option 4 ID: 49063914779 Status : Answered Chosen Option : ${\bf D}$ Marks: 1

Q.108 The gas that has relatively stronger affinity for Hemoglobin than Oxygen is: A. CO X B. NO₃ X c. CH₄ XD. NO2 Question Type: MCQ Question ID: 4906393710 Option 1 ID: 49063914840 Option 2 ID: 49063914838 Option 3 ID: 49063914839 Option 4 ID: 49063914837 Status : Answered Chosen Option: A Marks: 1 Q.109 The process that helps in nutrient conservation is: Ans X A. Leaching X B. Mineralization X C. Nitrification D. Immobilization Question Type: MCQ Question ID: 4906393700 Option 1 ID: 49063914800 Option 2 ID: 49063914797 Option 3 ID: 49063914799 Option 4 ID: 49063914798 Status : Answered Chosen Option : C Marks : -1/4 Q.110 Which among the following is a Renewable source of energy? Ans X A. Kerosene 🥒 B. Biomass X C. Coal X D. Petroleum Question Type : MCQQuestion ID: 4906393701 Option 1 ID: 49063914804 Option 2 ID: 49063914801 Option 3 ID: 49063914802 Option 4 ID: 49063914803 Status: Answered Chosen Option : ${\bf B}$ Marks: 1

Section: Chemistry

Q.111 Write IUPAC name of the given

Compound:

 $CH_2 = C (CH_2CH_2CH_3)_2$

Ans X A. 1-n-Propylpent-2-ene

X B. 2-n-Ethyldipropene

X C. Dipropylethene

Ø D. 2-n-Propylpent-1-ene

Question Type: MCQ

Question ID: 4906393738
Option 1 ID: 49063914949
Option 2 ID: 49063914952
Option 3 ID: 49063914950
Option 4 ID: 49063914951
Status: Answered

Chosen Option : **D** Marks : **1**

Q.112 What is the IUPAC name of the product obtained by addition reactions of HBr to Hex-1-ene in the presence of peroxide

Ans X A. 2-Bromohexane

X C. 5-Bromohexane

X D. 3-Bromohexane

Question Type: MCQ

Question ID: 4906393740
Option 1 ID: 49063914957
Option 2 ID: 49063914959
Option 3 ID: 49063914958
Option 4 ID: 49063914960
Status: Answered

Chosen Option : **B** Marks : **1**

Q.113 Which of the following species, do not show disproportionation reaction?

Ans

$$\mathscr{S}^{\text{B.}} \, ClO_4^-$$

Question Type: MCQ

Question ID: 4906393733
Option 1 ID: 49063914929
Option 2 ID: 49063914932
Option 3 ID: 49063914930
Option 4 ID: 49063914931
Status: Not Answered

Chosen Option : --

-- Marks : **0**

^{Q.114} For the equilibrium,

$$2NOCl(g) \rightleftharpoons 2NO(g) + Cl_2(g)$$

the value of the equilibrium constant, K_{c} is

 3.75×10^{-6} at 1069 K. Calculate the K_p for

the reaction at this temperature?

Ans X A. 0.003

✓ B. 0.033

X C. 0.33

X D. 3.33

Question Type: MCQ

Question ID: 4906393729
Option 1 ID: 49063914916
Option 2 ID: 49063914914
Option 3 ID: 49063914913
Option 4 ID: 49063914915
Status: Not Answered

Chosen Option : --

Q.115 What is the state of hybridisation of carbon in the given compound and its shape?

H₂C=O

s × A. sp³ hybridised carbon, tetrahedral

X C. sp hybridised carbon, linear

 \times D. sp^2 hybridised carbon, trigonal-pyramidal

Question Type: MCQ
Question ID: 4906393735
Option 1 ID: 49063914938
Option 2 ID: 49063914937
Option 3 ID: 49063914939
Option 4 ID: 49063914940
Status: Answered

Chosen Option : **B**Marks : **1**

^{Q.116} If N₂ gas is bubbled through water at 293K, how many millimoles of N₂ gas would dissolve in 1 litre of water? Assume that N₂ exerts a partial pressure of 0.987 bar. Given that Henry's law constant for N₂ at 293 K is 76.48 kbar.

X D. 1.432 mmol

Question Type: MCQ
Question ID: 4906393743
Option 1 ID: 49063914972
Option 2 ID: 49063914969
Option 3 ID: 49063914970
Option 4 ID: 49063914971
Status: Not Answered

Chosen Option : --Marks : 0

Q.117 What is the role of depressants in Froth floatation process?

A. It selectively prevents/allows elements to come with froth

X B. To enhance froth

💢 C. It makes the element heavier and help in settling below

X D. To reduce the amount of froth

Question Type: MCQ

Question ID: 4906393749 Option 1 ID: 49063914995 Option 2 ID: 49063914993 Option 3 ID: 49063914996 Option 4 ID: 49063914994 Status: Answered

Chosen Option: C Marks : -1/4

Q.118 How many moles of methane are required to produce 22g CO₂ (g) after combustion?

X A. 3 moles Ans

X B. 1 mole

X C. 2 moles

√ D. 0.5 mole

Question Type: MCQ

Question ID: 4906393712 Option 1 ID: 49063914848 Option 2 ID: 49063914846 Option 3 ID: 49063914845 Option 4 ID: 49063914847 Status: Answered

Chosen Option : D

Marks: 1

Q.119 Which of the following will have the least negative electron gain enthalpy?

Ans X A. F

X B.S

√ C. P

X D. CI

Question Type: MCQ

Question ID: 4906393722 Option 1 ID: 49063914888 Option 2 ID: 49063914886 Option 3 ID: 49063914885

Option 4 ID: 49063914887 Status: Answered

Chosen Option : C

Q.120 Although Thermodynamically feasible, but practically, Magnesium metal is not used for reduction of alumina in the Aluminium metallurgy. Why?

Ans X A. The pressure required is not attainable

💢 C. Magnesium cannot reduce Alumina in any conditions

X D. The temperature required is not attainable

Question Type: MCQ

Question ID: 4906393748
Option 1 ID: 49063914992
Option 2 ID: 49063914990
Option 3 ID: 49063914989
Option 4 ID: 49063914991
Status: Answered

Chosen Option : C Marks : -1/4

Q.121 The vapour pressure of pure benzene at a certain temperature is 0.850 bar. A non-volatile, non-electrolyte solid weighing 0.5 g when added to 39.0 g of benzene (molar mass 78 g mol-1). Vapour pressure of the solution, then, is 0.845 bar. What is the molar mass of the solid substance?

Ans

× 17 g mol⁻¹

è 170 g mol⁻¹

× c. 85 g mol⁻¹

× • 340 g mol⁻¹

Question Type : MCQ

Question ID: 4906393744
Option 1 ID: 49063914973
Option 2 ID: 49063914976
Option 3 ID: 49063914974
Option 4 ID: 49063914975
Status: Not Answered

Chosen Option : --Marks : 0 Q.122 The number of electrons, protons and neutrons in a species are equal to 18, 16 and 16 respectively. Then, the proper symbol of the species would be:

X A. S2+

× B. C|2-× C. P2-✓ D. S2-

Question Type: MCQ

Question ID: 4906393719 Option 1 ID: 49063914873 Option 2 ID: 49063914875 Option 3 ID: 49063914874 Option 4 ID: 49063914876

Status: Answered Chosen Option : D

Q.123

The pK_a of acetic acid and pK_b of ammonium hydroxide are 4.76 and 4.75 respectively. Calculate the pH of ammonium. acetate solution.

X A. 0.0075 Ans

X B. 5.007

√ C. 7.005

X D. 0.757

Question Type : MCQ

Question ID: 4906393731 Option 1 ID: 49063914921 Option 2 ID: 49063914923 Option 3 ID: 49063914922 Option 4 ID: 49063914924 Status: Answered

Chosen Option: C Marks: 1

Q.124

The boiling point of benzene is 353.23 K. When 1.80 g of a non-volatile solute was dissolved in 90 g of benzene, the boiling point is raised to 354.11 K. Calculate the molar mass of the solute.

[K_b for benzene is 2.53 K kg mol⁻¹]

Ans

- × 116 g mol⁻¹
- ×в. 98 g mol⁻¹
- √ °. 58 g mol⁻¹

Question Type: MCQ

Question ID: 4906393745 Option 1 ID: 49063914979 Option 2 ID: 49063914980

Option 3 ID: 49063914978 Option 4 ID: 49063914977

Status : Not Answered

Chosen Option : --Marks : 0

Q.125 Which of the following correctly depicts the repulsive interaction of Lone pair (lp) and Bond pair (bp) of electrons?

Ans X A. (lp-bp) > (bp-bp) > (lp-lp)

✗ B. (lp-bp) > (lp-lp) > (lp-bp)

C. (lp-lp) > (lp-bp) > (bp-bp)

X D. (lp-lp) > (bp-bp) > (lp-bp)

Question Type: MCQ

Question ID : 4906393727

Option 1 ID: 49063914906

Option 2 ID : 49063914908

Option 3 ID: 49063914907

Option 4 ID: 49063914905

Status : **Answered**

Chosen Option : C

Q.126 What would be the IUPAC Nomenclature for the element with atomic number 110? Ans X A. Unbinilium B. Ununnilium C. Unnilunium D. Ununbium

Question Type: MCQ
 Question ID: 4906393718
 Option 1 ID: 49063914870
 Option 2 ID: 49063914872
 Option 3 ID: 49063914871
 Option 4 ID: 49063914869
 Status: Answered
Chosen Option: B
 Marks: 1

Q.127 45 g of ethylene glycol (C₂H₆O₂) is mixed with 600 g of water. Calculate the freezing point of the solution?

[K_f for water is 1.86 K kg mol⁻¹]

Ans X A. 280.56 K

X B. 273.202 K

√ C. 270.95 K

X D. 265.67 K

Question Type : \mathbf{MCQ}

Question ID: 4906393746 Option 1 ID: 49063914983 Option 2 ID: 49063914981

Option 3 ID : **49063914982** Option 4 ID : **49063914984**

Status : Not Answered

Chosen Option : --

Q.128

50.0 kg of N_2 (g) and 10.0 kg of H_2 (g) are mixed to produce NH_3 (g).

$$N_2(g) + 3 H_2(g) \rightleftharpoons 2NH_3(g)$$

Which is the limiting reagent in the production of NH₃ in the above case?

Ans

XA both N2 and H2

X B. None of them

X c. N2

√ D. H₂

Question Type: MCQ

Question ID: 4906393713

Option 1 ID: 49063914851

Option 2 ID: 49063914852

Option 3 ID: 49063914849 Option 4 ID: 49063914850

Status : Answered

Chosen Option : **D**

Marks: 1

Q.129 Which of the following is a Nucleophile?

Ans X A. N⁺O₂

X B. CH₃-C⁺=O,

X c. Cl*

⊘ D. HS⁻

Question Type: MCQ

Question ID: 4906393736

Option 1 ID: 49063914944

Option 2 ID: 49063914943

Option 3 ID: 49063914942

Option 4 ID: 49063914941

Status: Answered

Chosen Option : D

 $^{ extsf{Q.130}}$ How many σ and π bonds are present in the given molecule?

HC≡CCH=CHCH₃

Ans \times A. 9 σ and 4 π bonds

 \nearrow B. 12 σ and 2 π bonds

 χ C. 11 σ and 2 π bonds

 $\checkmark\!\!\!/$ D. 10 σ and 3 π bonds

Question Type: MCQ

Question ID: 4906393734 Option 1 ID: 49063914935 Option 2 ID: 49063914936

Option 3 ID: 49063914933 Option 4 ID: 49063914934

Status : Answered Chosen Option : D

Marks: 1

Q.131 Which of the following species will have the smallest size?

Ans

XA.AI

×∘ Mg

× □. Mg²⁺

Question Type: MCQ

Question ID: 4906393721 Option 1 ID: 49063914883

Option 2 ID: 49063914884 Option 3 ID: 49063914881

Option 4 ID: 49063914882 Status : Answered

Chosen Option : B

Q.132 At equilibrium, the concentrations of

$$N_2$$
=3.0 × 10⁻³M,

$$O_2 = 4.2 \times 10^{-3} M$$
 and

$$NO = 2.8 \times 10^{-3}M$$

in a sealed vessel at 800K. What will be K_c for the reaction?

$$N_2(g) + O_2(g) \rightleftharpoons 2NO(g)$$

Ans X A. 0.226

X B. 6.22

√ C. 0.622

X D. 622

Question Type: MCQ

Question ID: 4906393728

Option 1 ID: 49063914912

Option 2 ID: 49063914910

Option 3 ID: 49063914909

Option 4 ID: 49063914911 Status: Answered

Chosen Option : C

Marks : 1

Q.133 Using the Periodic Table, predict the formulas of compounds which might be formed by the following pairs of elements: aluminium and sulphur.

Ans

Question Type: MCQ

Question ID: 4906393723

Option 1 ID: 49063914889

Option 2 ID: 49063914890

Option 3 ID: 49063914891

Option 4 ID: 49063914892

Status : Answered

Chosen Option : C

Q.134 A golf ball has a mass of 40g, and a speed of 45 m/s. If the speed can be measured within accuracy of 2%, calculate the uncertainty in the position?

Ans

Question Type: MCQ

Question ID: 4906393716 Option 1 ID: 49063914864 Option 2 ID: 49063914863 Option 3 ID: 49063914862

Option 4 ID: 49063914861 Status: Not Answered

Chosen Option : --

Marks: 0

Q.135 Which of the following is not a method of Concentration of Ores?

Ans X A. Leaching

X B. Magnetic separation

X C. Froth Floatation

🦪 D. Liquation

Question Type: MCQ

Question ID: 4906393750 Option 1 ID: 49063915000 Option 2 ID: 49063914998 Option 3 ID: 49063914997

Option 4 ID: 49063914999

Status : Answered

Chosen Option : D

Q.136 Calculate the molarity of a solution containing 5 g of NaOH in 450 mL solution?

Ans

 \times 0.728 mol L⁻¹

√ B. 0.278 mol L⁻¹

× c. 0.872 mol L⁻¹

× 0.556 mol L⁻¹

Question Type: MCQ

Question ID: 4906393742
Option 1 ID: 49063914965
Option 2 ID: 49063914967
Option 3 ID: 49063914966
Option 4 ID: 49063914968
Status: Answered

Chosen Option : **B**Marks : **1**

Q.137 A 100 watt bulb emits monochromatic light of wavelength 400 nm. Calculate the number of photons emitted per second by the bulb?

Ans

 \times 9.078 X 10^{20}

×в. 8.412 X 10²⁵

× c. 4.12 X 10³⁰

√ □ 2.012 X 10²⁰

Question Type : \mathbf{MCQ}

Question ID: 4906393714 Option 1 ID: 49063914856 Option 2 ID: 49063914855 Option 3 ID: 49063914853 Option 4 ID: 49063914854 Status: Answered

Chosen Option : C Marks : -1/4 Q.138 Which of the following compounds will show cis-trans isomerism?

(i) $(CH_3)_2C = CH - C_2H_5$

(ii) $CH_2 = CBr_2$

(iii) $C_6H_5CH = CH - CH_3$

(iv) CH₃CH = CCI CH₃

Ans 🗶 A. only (ii)

X C. (i), (ii), and (iii) only

X D. only (i) and (ii)

Question Type: MCQ

Question ID: 4906393739

Option 1 ID: 49063914953

Option 2 ID: 49063914956

Option 3 ID: 49063914955

Option 4 ID: 49063914954

Status : Answered

Chosen Option : A

Marks : -1/4

200 cm³ of an aqueous solution of a protein contains 1.26 g of the protein. The osmotic pressure of such a solution at 300 K is found to be 2.57 × 10⁻³ bar. Calculate the molar mass of the protein?

Ans

- * 15,346 g mol⁻¹
- × 8 30,998 g mol⁻¹
- × 126,065 g mol⁻¹

Question Type: MCQ

Question ID: 4906393747 Option 1 ID: 49063914988 Option 2 ID: 49063914986

Option 3 ID: 49063914985 Option 4 ID: 49063914987

Status : Answered

Chosen Option : B Marks : -1/4

Q.140 Which of the following could be true about the factors affecting equilibria?

Ans X A. The concentration stress of a removed reactant/product is relieved by net reaction in the direction that reduces the removed substance.

X B. The equilibrium constant for an endothermic reaction decreases as the temperature increases.

C. Catalyst does not affect the equilibrium composition of a reaction mixture.

X D. The equilibrium constant for an exothermic reaction increases as the temperature increases.

Question Type : \mathbf{MCQ}

Question ID: 4906393730

Option 1 ID : 49063914917

Option 2 ID: 49063914919

Option 3 ID : 49063914920

Option 4 ID : 49063914918

Status : Answered

Chosen Option : C

Q.141

The values of K_{sp} of two sparingly soluble salts Ni(OH)₂ and AgCN are 2.0 \times 10⁻¹⁵ and

 6×10^{-17} respectively. Which salt is more

X A. cannot be determined Ans

X B. both are equally soluble

X C. AgCN

√ D. Ni(OH)₂

Question Type: MCQ

Question ID: 4906393732

Option 1 ID: 49063914928

Option 2 ID: 49063914927

Option 3 ID: 49063914926

Option 4 ID: 49063914925

Status: Not Answered

Chosen Option: --

Marks: 0

Q.142 Calculate the mass of a photon with wavelength 3.6 A.

× 16.135 x 10-29 kg

×в. 8.139 x 10-29 kg

Question Type: MCQ

Question ID: 4906393715

Option 1 ID: 49063914859

Option 2 ID: 49063914860 Option 3 ID: 49063914857

Option 4 ID: 49063914858

Status: Answered

Chosen Option: C

Q.143 Using s, p, d, f notations, describe the orbital with the following quantum numbers: n = 5, l = 3 Ans A. 5f X B. 5d X C. 5s X D. 5p Question Type: MCQ Question ID: 4906393717 Option 1 ID: 49063914868 Option 2 ID: 49063914865 Option 3 ID: 49063914866 Option 4 ID: 49063914867 Status : Answered Chosen Option: A Marks: 1 Q.144 On complete combustion, 0.246 g of an organic compound gave 0.198g of carbon dioxide and 0.1014g of water. Determine the percentage composition of Hydrogen in the compound? ✓ A. 0.0458 Ans X B. 0.1066 X C. 0.0936 X D. 0.0688 Question Type : MCQQuestion ID: 4906393737 Option 1 ID: 49063914946 Option 2 ID: 49063914948 Option 3 ID: 49063914945 Option 4 ID: 49063914947 Status: Not Answered Chosen Option: --Marks: 0 Q.145 Considering the atomic number and position in the periodic table, arrange the following elements in the increasing order of metallic character: Si, Be, Mg, Na, P. Ans X B. Si < P < Be < Na < Mg X C. Si < Be < P < Na < Mg X D. P < Si < Be < Na < Mg Question Type: MCQ Question ID: 4906393720 Option 1 ID: 49063914877 Option 2 ID: 49063914879 Option 3 ID: 49063914880 Option 4 ID: 49063914878 Status: AnsweredChosen Option: A Marks: 1

Calculate the approximate molecular mass of glucose (C₆H₁₂O₆) molecule ?

Ans X A. 198 u

X B. 150 u

X C. 168 u

√ D. 180 u

Question Type: MCQ

Question ID: 4906393711
Option 1 ID: 49063914844
Option 2 ID: 49063914841
Option 3 ID: 49063914843
Option 4 ID: 49063914842
Status: Answered

Chosen Option : **D**Marks : **1**

L___

Q.147

In which shaped molecules/ions sp^3d^2 hybridization can happen?

Ans X A. None of these

X B. square-pyramidal

X C. octahedral

 \mathscr{P} D. both square-pyramidal and octahedral

Question Type: MCQ

Question ID: 4906393726
Option 1 ID: 49063914904
Option 2 ID: 49063914901
Option 3 ID: 49063914902
Option 4 ID: 49063914903
Status: Answered

Chosen Option : **D** Marks : **1** Q.148 Calculate the mole fraction of water in a solution containing 20% of C₂H₆O₂ by mass. Ans X A. 0.438 X B. 0.876 X C. 0.068 √ D. 0.932 Question Type: MCQ Question ID: 4906393741 Option 1 ID: 49063914964 Option 2 ID: 49063914963 Option 3 ID: 49063914962 Option 4 ID: 49063914961 Status: Not Answered Chosen Option: --Marks: 0 Q.149 What is the covalency of AI in AICI(H_2O)₅]²⁺ ? √ A. 6 Ans **X** B. 5 X C. 3 X D. 4 Question Type : \mathbf{MCQ} Question ID: 4906393724 Option 1 ID: 49063914896 Option 2 ID: 49063914895 Option 3 ID: 49063914893 Option 4 ID: 49063914894 Status : Answered Chosen Option : C Marks : -1/4

which of the following is the geometry of NF3 molecule?

Ans 🗶 A. Trigonal-planar

X B. Tetrahedral

X C. Bent

D. Trigonal-Pyramidal

Question Type : \mathbf{MCQ}

Question ID: 4906393725
Option 1 ID: 49063914900
Option 2 ID: 49063914898
Option 3 ID: 49063914899
Option 4 ID: 49063914897
Status: Answered

Chosen Option : **D** Marks : **1**